

News & Views

Long-Lasting Antioxidant Activity in a 600-Year-Old Fermented Fruit Juice

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WINE-BASED MEDICAL THERAPEUTICS

MODERN MEDICINE has recently acknowledged the therapeutic value of an ancient remedy: wine.

Wine, together with olive oil, fruit, vegetables, cereals, and fish, is one of the staples of the so-called Mediterranean diet, which was found to be associated with low cardiovascular mortality more than 50 years ago (16). More recently, epidemiological studies have shown that moderate consumption of wine, but not of alcohol, is associated with a reduction in cardiovascular mortality (15, 21, 24, 27, 28, 38, 46). A meta-analysis of 13 studies in nearly 300,000 subjects has confirmed that moderate wine consumption has cardioprotective properties and reduces also the risk of cerebrovascular disease (18).

The constituents of wine differ considerably from those of other alcoholic beverages (9), and many experimental studies have been carried out to identify its active principles.

Antioxidants are considered to be responsible for the protective effects of the Mediterranean diet, together with its high content in polyunsaturated fats (16, 48).

Wine, especially red wine, contains considerable quantities of polyphenols (up to 5 g/L), which are well-known antioxidants and free radical scavengers. *In vitro* studies have shown that these polyphenols are able to inhibit human low-density lipoprotein oxidation and could therefore contribute toward the prevention of thrombosis and atherosclerosis (20, 46).

This hypothesis is supported by the recent finding of an inverse relationship between oxidized low-density lipoprotein antibodies and daily wine consumption in a cross-sectional study in 551 elderly Italian subjects (17). Moreover, the scavenging properties of polyphenols can inhibit the interaction between endothelium-derived nitric oxide (NO) and superoxide anion. This increases NO levels in the endothelium and may contribute toward prevention of endothelial dysfunction, which occurs in the early stages of atherogenesis and promotes thrombogenesis (46).

The positive effects of wine on endothelial dysfunction are supported by a study in 42 healthy volunteers allocated either to a Mediterranean diet or to a high-fat diet for 1 month and then given wine (240 ml daily) for another month. Wine was able to correct the reduction in brachial artery diameter observed in the subjects on the high-fat diet; it also enhanced the positive effects of the Mediterranean diet and corrected the negative effects of a high-fat diet on total plasma antioxidant capacity (29).

Resveratrol is one of the most important typical polyphenols of wine. In the hearts of mice subjected to ischemia and reperfusion, it reduced myocardial infarct size and the number of apoptotic cardiomyocytes, and improved postischemic ventricular function to a significantly greater extent than no treatment or ethanol. Moreover, the lower malonaldehyde content of the hearts of resveratrol-treated mice showed that oxidative stress was lower in these animals than in controls (42). These properties were no longer evident in inducible NO synthase knockout mice or in the heart of wild-type mice given an inducible NO synthase inhibitor (26). Thus, NO up-regulation and reduction of oxidative stress appear to be the key cardioprotective mechanisms.

Another important mechanism that has recently been shown for polyphenols contained in olive oil and red wine, including resveratrol, is the transcriptional inhibition of vascular cell adhesion molecule-1 (VCAM-1); this finding indicates that the Mediterranean diet prevents endothelial leukocyte adhesion, an early step in atherogenesis (10, 19). The results of these studies also suggest that a Mediterranean diet may modify familial predisposition to cardiovascular disease by regulating the expression on proinflammatory/ proatherogenic genes.

Resveratrol also possesses other biological properties of interest: antiinflammatory activity, expressed as reduction in the production of tumor necrosis factor- α (TNF- α), interleukin-1 β , interleukin-8, and granulocyte macrophage colony stimulating factor, antiinfective activity, including

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stimulation of phagocytosis, and anticancer activity, including cytostatic activity, stimulation of apoptosis, and antimutagenic effects (4, 6, 7, 14).

The chemical composition of the various types of wine can differ considerably. Major differences have been ascertained between red and white wine, the latter being poor in resveratrol and in polyphenols, in general.

Nevertheless, an ethanol-free white wine extract was able to exert cardioprotective effects: it increased aortic and coronary blood flow, as well as left ventricular developed pressure, and reduced myocardial infarct size, as well as malondialdehyde, in rat hearts significantly versus controls and other ethanol-free white wine extracts. Interestingly, the cardioprotective white wine extract exerted the most pronounced antioxidant activity (13). The monophenols tyrosol and caffeic acid, which are present in extra virgin olive oil, appear to be the main compounds that contribute to the antioxidant activity of white wine. These compounds also exert other effects, such as the inhibition of the release of inflammatory cytokines, including TNF- α , interleukin-1 β , and interleukin-6, at concentrations that are found in the bloodstream after modest wine or olive oil intake (3, 5, 22).

Also grape extracts and seeds have proved to be cardioprotective in the same heart model used for the studies with red and white wine (12, 35, 43). Grapes also contain melanin, which is an antioxidant that possesses antiinflammatory and immunomodulating properties, including the ability to inhibit cytokine (TNF- α , interleukin-1 β , and interleukin-6) production (2).

WINE AS AN ARCHAEOLOGICAL ARTEFACT

Technological advances enable investigators to carry out in-depth investigations on the composition of very small traces of substances on archaeological artefacts. The detection of large quantities of tartaric acid identifies the trace as the remnant of grape juice or wine; the detection of polyphenols and of fermentation derivatives is then required for the differential diagnosis (31). This can easily be achieved with high-pressure liquid chromatography (HPLC) and gas chromatography (GC).

Archaeochemists have thus found traces of wine on a number of artefacts, including a 7,000-year-old Neolithic jar, which pushes the history of wine making back by 2,000 years (31).

A number of authors have provided us with literature on wine-making techniques and usage in Greek and Roman times, when wine played an important role in medicine: it was the basic ingredient of dozens of "composite" remedies including also other ingredients, such as fruit juice, spices, roots, and honey (49). The historian Pliny the Elder (37) tells us that wine was always diluted with something else, such as water, which could also be seawater "to enliven the wine's smoothness." Columella (11) describes the procedure of simmering grape syrup slowly in a lead pot or lead-lined copper kettle; this resulted in the formation of "sugar of lead" or lead acetate, a potent fungicide that preserved the fluid well. In the last century, Hoffman followed the recipes of Columella and

ended up with wines that contained 15–30 mg/L lead acetate—50–100 times the currently legal limit (32, 33). Indeed, Pliny describes a pandemic disease that resembles saturnine gout; historians claim that the disease contributed to the fall of the Roman empire (32, 37).

The medicinal use of wine continued during the Middle Ages, when its virtues were extolled not only by the great school of medicine at Salerno (Italy), but even by textbooks of Arabic medicine. At the same time, the Chinese used plant extracts containing high concentrations of polyphenols, such as resveratrol (49).

THE DISCOVERY AT FAVIGNANA

In September 2000, the corroded relict of a ship was found offshore along the west coast of Sicily, at Favignana, near Marsala, by Sicily's regional archaeological team, led by Sebastiano Tusa (44) (Fig. 1). Its cargo consisted of a number of unusually small amphorae (27.6 cm high including the stopper) made of pewter and of notable artistic value (Fig. 2). Subsequent archaeological investigations established that the vessels dated back to 1300–1400 AD.

The amphorae were tightly closed with screw stoppers, and the contents of some of them were still intact and liquid. The small size and artistic beauty of the vessels induced us to speculate that they contained a special liquid, possibly a medicinal remedy, and to ask the archaeologists to give us a sample so we could analyze it.

WHAT WAS THE MEDIEVAL FLUID FOUND IN THE FAVIGNANA AMPHORAE?

The medieval fluid was submitted to HPLC (see Appendix, note 1). Following detection of tartaric acid, the experimental plan was laid for the detection of organic acids and phenols usually present in compounds belonging to the vegetable kingdom (see Appendix, note 2). The results are shown in Table 1. The finding of tartaric acid indicated that the medieval fluid was a fruit juice, made at least partly from grapes. The concentration of ascorbic acid (1.6 mg/ml) indicated that fruit juice rich in vitamin C, such as orange juice, had been added, as the content of ascorbic acid in grapes is low (36).

The medieval fluid was then submitted to GC focused on the detection of primary and secondary compounds resulting from the fermentation of grapes (see Appendix, note 3). This investigation detected a low concentration of ethanol [3.01 g/100 ml, three to four times lower than the concentration in modern wine (40)] and compounds that are the typical result of secondary fermentation of grapes: acetaldehyde, methanol, 2-methyl-1-butanol and 3-methyl-1-butanol (39) (Table 2).

The medieval fluid was subsequently subjected to atomic absorption spectrophotometry to establish its content in sodium and lead (see Appendix, note 4). The Na⁺ concentration was low (4,608 ppm). As sea infiltration was a possible explanation for the low concentration of ethanol, a comparison was made with threefold-diluted seawater samples collected near the site of wreckage. The concentrations results were similar:



FIG. 1. Site of wreck in the Mediterranean Sea.

medieval fluid 4,608 versus 4,900 ppm, suggesting that sea-water had seeped into the vessel through microcracks, diluting the fluid. However, the intentional addition of salty water to the juice “to enliven the wine’s smoothness” as described by Pliny (37) cannot be ruled out.

The concentration of lead was found to be high (6 mg/L, equivalent to 20 times the current legal limit of 0.3 mg/L) (33). The medieval fluid was contained in amphorae made of pewter, an alloy that contains lead, so its high concentration

may be due to leaching from the vessel for centuries. However, it is known that Roman food and wine were contaminated by lead following the custom to simmer grape syrup in lead pots, and lead poisoning is believed to have contributed to the fall of the Roman Empire (11, 32). Thus, the high concentration may have been due to lead contamination during its production, to which leaching contributed even further. Either way, the finding of a fluid for oral consumption in a vessel partly made of lead suggests that lead poisoning had still not been recognized in the Middle Ages. Indeed, lead-induced gout, an inflammatory disorder involving the joints, was widespread in the upper classes in England three centuries later (1).

TABLE 1. HPLC ANALYSIS OF MEDIEVAL FLUID

	Retention Time (min)	Concentration (mg/ml)
$\lambda = 210 \text{ nm}$		
Chinic acid	5.3	64.0
Tartaric acid	5.6	1.2
Malic acid	7.2	0.6
Lactic acid	8.7	44.1
Acetic acid	9.1	78.5
Succinic acid	15.4	60.6
$\lambda = 254 \text{ nm}$		
Ascorbic acid	8.9	1.6

TABLE 2. GC ANALYSIS OF MEDIEVAL FLUID: SECONDARY FERMENTATION COMPOUNDS

Substance	Concentration (mg/L)
Acetaldehyde	65
Methanol	20
2-Methyl-1-butanol	15
3-Methyl-1-butanol	69



FIG. 2. Photograph of sealed pewter amphora shortly after retrieval from the shipwreck.

The next investigation was aimed at the quantification of phenols expressed by the Folin-Ciocalteu index (45), using wines as reference fluids (see Appendix, note 5). The polyphenol content of the medieval fluid was low as compared with the red wine and white Marsala wine: the Folin-Ciocalteu polyphenol index was only 9 mg of gallic acid/100 g versus 285 mg/100 g in the red wine and 32 mg/100 g in the white Marsala wine.

In summary, the analysis of the medieval fluid with modern technology established that it was a fermented white grape juice, to which citrus juice and salty water may have been added; the fluid was contaminated with lead.

THE ANTIOXIDANT ACTIVITY OF THE MEDIEVAL FLUID

The antioxidant activity of the medieval fluid was investigated by radical 1,1-diphenyl-2-picrylhydrazyl (DPPH) detection and electron paramagnetic resonance (EPR) and compared with that of modern products of grape fermentation, *i.e.*, a modern threefold-diluted red wine and white Marsala wine.

Radical DPPH detection, a spectrophotometric test that evaluates antioxidant activity, was performed according to the

method of Brand-Williams *et al.* (8) with a few modifications (see Appendix, note 6).

Antioxidant activity was also measured against the hydroxyl (HOH) free radical generated in a Fenton system by using an EPR spin-trapping method (41) (see Appendix, note 7).

According to DPPH, red wine exerted the highest antioxidant activity, whereas the antioxidant activities of the medieval fluid and white Marsala wine were almost identical (Fig. 3).

The similarity of the antioxidant activities of the medieval fluid and the white Marsala wine was confirmed by EPR already after twofold concentration of the medieval fluid (scavenging activity: red wine, 100.0; white Marsala wine, 81.7; medieval fluid, 48.5; medieval fluid concentrated 2:1, 82.8).

OTHER BIOLOGICAL PROPERTIES OF THE MEDIEVAL FLUID

The immunomodulatory and antiinflammatory activities of the medieval fluid were studied, because substances contained in fermented grape juice, such as resveratrol, melanin, tyrosol, caffeic acid, and others, possess these properties (2–4, 6, 12–14, 22, 35, 43).

The immunomodulatory activity of the medieval fluid was studied by measuring the capacity of U937 human monocyte cells obtained from the American Tissue Culture Collection (Rockville, MD, U.S.A.) to phagocytize *Bacillus subtilis* after incubation with the medieval fluid in the presence and absence of lead versus incubation with a control solution containing 3% alcohol and salt. The method has already been described in detail elsewhere (50).

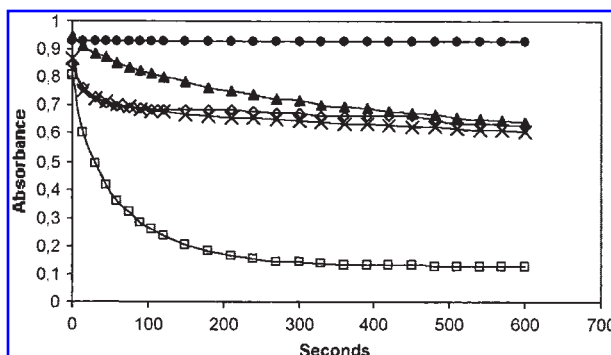


FIG. 3. Results of antioxidant assay using the stable synthetic free radical DPPH. Antioxidant activity was measured by evaluating absorbance decrease at 517 nm in the presence of potential scavenger products, and comparing results with a well-known scavenger molecule, Trolox. A reference solution was made by adding 0.1 ml of a 1 mM ethanolic Trolox solution to the reaction mixture (final concentration 0.04 mM). Threefold-diluted red wine (\square) produced a clear reduction, reaching a plateau at ~ 0.2 absorbance units 200 s after the reaction. Plots of medieval fluid (\times) and white wine (\blacktriangle) were close to the plot of 0.013 mM Trolox (\blacklozenge), especially at the end of the reaction time, but white wine initially had slower kinetics of DPPH degradation than medieval fluid. \bullet , reference.

The antiinflammatory activity of the medieval fluid was studied measuring the levels of inducible TNF- α , which has proved to play an important role in inflammatory diseases, such as rheumatoid arthritis and ankylosing spondylitis, so that its inhibition may be considered a clinically meaningful index of antiinflammatory properties (23, 34). TNF- α levels were measured in the supernatants collected from the samples of fluid from the phagocytosis experiment. The method has already been described in detail elsewhere (5).

The medieval fluid reduced the release of TNF- α by human monocytes independently of lead (medieval fluid - Pb, $2,618 \pm 0.18$ pg/ml; medieval fluid + Pb, $2,381 \pm 0.31$ pg/ml; versus 3% alcohol + salt control, $17,739 \pm 5,485$ pg/ml), but increased their capacity to phagocytize *Bacillus subtilis* only after lead removal (phagocytosis frequency: medieval fluid - Pb = $62 \pm 31\%$, medieval fluid + Pb = $43 \pm 26\%$; versus 3% alcohol + salt control = $46 \pm 14\%$). The results confirming nonspecific immunity enhancement are consistent with previous data obtained with resveratrol, tyrosol, and melanin, and with Roman medical practice (Galen used wine to disinfect the wounds of gladiators) (2–4, 14, 22, 49).

WHAT HAVE WE LEARNED FROM THE ANALYSIS OF A 600-YEAR-OLD FRUIT JUICE?

To our knowledge, this is the first time that the antioxidant activity and other biological properties of a medieval fruit juice have been studied with modern techniques.

The conclusions that can be drawn from this observational report are limited by the small sample size due to the small amount of medieval fluid made available to us. However, all differences were important from a biological point of view (Fig. 3).

The very long-term persistence of antioxidant activity of fermented fruit juice stresses the importance of fruit in maintaining the oxidant/antioxidant balance in the organism and its role as a staple of the Mediterranean diet.

An open issue is what was responsible for the persistent antioxidant activity. Both red and white wines notoriously possess antioxidant activity thanks to their content of poly- and monophenols (13, 15), which were scanty in the medieval fluid. However, it did contain a fairly high concentration of another antioxidant, namely, ascorbic acid (1.6 mg/ml), which therefore appears to have contributed to the positive outcome of the tests.

Interestingly, modern technology established that the composition of the medieval fluid corresponded to recipes for "composite" medicinal remedies used in Roman times: wine (the fermented grape juice may have been wine, although this cannot be demonstrated with certainty) diluted with seawater and citrus fruit juice. Thus, the hypothesis that the small pewter vessel contained a medicinal remedy appears to have been correct.

The sea bottom environment reproduces the conditions of the ideal wine cellar: total darkness, which prevents light from giving wine a cardboard-type flavor, and stable temperature only a few degrees above freezing, which slows down the aging process (25, 47). Indeed, century-old champagne has been retrieved from shipwrecks in excellent condition,

when the bottle was properly sealed as the medieval pewter vessel was (30, 47).

What did the sommelier have to say about the medieval fermented grape juice after 600 years? "Dull white color with a pale golden hue; aroma of leather-dried grapes and aromatic vine; acid, astringent and salty taste devoid of bitterness. The taste is strong and unusually persistent . . ." Tasting was not allowed to continue in view of the high lead content of the fluid.

In conclusion, the opportunity to analyze a 600-year-old fruit juice has provided evidence for the first time that the antioxidant, antiinflammatory, and immunomodulatory properties of fermented fruit can persist for centuries, provided that it is properly stored.

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ABBREVIATIONS

DMPO, 5,5-dimethylpyrrolidine-*N*-oxide; DPPH, 1,1-diphenyl-2-picrylhydrazyl; EPR, electron paramagnetic resonance; GC, gas chromatography; HPLC, high-pressure liquid chromatography; NO, nitric oxide; \cdot OH, hydroxyl radical; PBS, phosphate buffer solution; TNF- α , tumor necrosis factor- α .

APPENDIX

1. Inertsil ODS-3 4.6×250 mm column with 5 μ m particle size and H_3PO_4 0.02 *M* mobile phase at a flow rate of 0.7 ml/min at 20°C; repeat analysis: Inertsil PH 4.6×250 mm column with 5 μ m particle size and $\text{H}_2\text{O}/\text{CH}_3\text{CN}/\text{CH}_3\text{COOH}$ (80:17:3) mobile phase at a flow rate of 0.5 ml/min at 50°C; extraction protocol: centrifugation and fivefold dilution with mobile phase.
2. Compounds were detected by an ultraviolet wave detector at 210 (organic acids, including acetic, succinic, lactic, malic, and ascorbic acids), 254 (ascorbic acid), 280 (simple phenols), and 320 (cinnamates) nm.
3. Headspace method, using an automatic headspace sampler, conditioning 2 ml of each sample in a closed 3-ml vial at 80°C for 1 hr; 500 μ l of headspace was injected into the gas chromatograph; a 30-m Carbowax 20M column was used with He as carrier at 0.8 ml/min, programmed at 40°C for 5 min, then 30°C/min and 150°C for 5 min; the injector was at 230°C, FID was at 250°C; retention time of ethanol was 5.2 min.
4. PerkinElmer atomic spectrophotometer model M2100, sodium assay on the substrates (0.4 ml) dissolved in 3.6 ml of HNO_3 0.1 *M* and $\text{La}(\text{NO}_3)_3 \cdot 6\text{H}_2\text{O}$ 5 mM after calibration

with standards; lead was determined by graphite furnace atomic absorption spectrometry.

5. Method: the following were poured into a 25-ml flask: 0.1 ml of water (blank) or 0.1 ml of 0.1% gallic acid solution in water (reference) or 0.1-ml sample of medieval fluid or one of two different kinds of wine, a threefold diluted red wine and white Marsala wine; 1.25 ml of Folin–Ciocalteu reagent; 2.50 ml ml Na_2CO_3 -saturated solution; these solutions were set to 25-ml volume with water, kept 2 h in the dark, and their absorbance was read against blank at 730 nm. Phenol index was expressed as mg of gallic acid/100 ml of solution.
6. A final reaction volume of 2.6 ml was used. The assayed substrates (medieval fluid or modern wines) were added (0.1 ml) to 96% spectrophotometric grade ethanol (1.9 ml) and 0.1 ml of H_2O_2 , then were mixed with the DPPH solution (0.5 ml, 0.5 mM) to the final concentration of 0.1 mM. The blank solution was made by adding the same volume (0.1 ml) of pure ethanol solution to the DPPH solution; a reference solution was made by adding 0.1 mM ethanol Trolox solution (final concentration 0.04 mM) to the reaction mixture. The kinetics of DPPH bleaching was calculated by reading the absorbance at 517 nm every 30 s for 10 min, immediately after mixing the solutions. Each assay was replicated five times.
7. The hydroxyl radical ($\cdot\text{OH}$) produced in the Fenton reaction mixture was trapped with 5,5-dimethylpyrrolidine-*N*-oxide (DMPO). The resultant adduct DMPO $\cdot\text{OH}$, consisting of four vertical signals, was detected by an X-band EPR spectrometer Varian E-line Century series. For $\cdot\text{OH}$ measurement, the mixture without the scavenging compounds (blank) contained: 0.7 of phosphate buffer solution, pH 7.4 (PBS); 0.2 ml of Fe-EDTA 10–12 mM; 0.2 ml of 50 mM DMPO 50 mM in PBS; 0.2 ml of H_2O_2 10 mM. The mixture with the test scavenging compound contained: 0.5 ml of PBS; 0.2 ml of scavenger (wine or diluted wine); 0.2 ml of Fe-EDTA 10–12 mM; 0.2 ml of 50 mM DMPO 50 mM in PBS; 0.2 ml of H_2O_2 10 mM. These solutions were accurately mixed in a glass tube assay and successively placed in the EPR probe, a capillary tube of 100-mm length and 1.3-mm internal diameter. EPR spectra were recorded after exactly 2 min. The instrumental parameters were: frequency, 9.26 GHz; power, 5 mW; field set, 3,390 gauss; scan time, 64 s; time constant, 0.5 s; gain, 16,000; modulation, 1 gauss. The scavenger activity percent of the test compound for ^1OH was expressed by the following formula: $I = 100 - [(h_x/h_0)] \times 100$ where I was scavenger activity, h_0 and h_x were the relative heights of the signal 2 (mm) of the DMPO- ^1OH adduct spectra in a reaction mixture without and with the scavenger compound, respectively. To avoid interference of ethanol in the EPR analysis, the wines were deprived of alcohol at low temperature in a centrifugal evaporator; the initial volume was restored with water. Each assay was repeated 10 times.

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